

## PHOSPHORUS IMMOBILIZING AMENDMENTS TO SOIL

2011

authors: W.J. Chardon and J.M. Dorioz

### *Description*

The solubility of phosphorus in the soil is reduced by adding a substrate that contains P sorbing compounds, with the aim of reducing the risk of dissolved P losses.

### *Rationale, mechanism of action*

When the P content of a soil is high this can lead to large losses of dissolved P to surface or ground water. Lowering the solubility of P by adding a substrate that contain a large amount of iron (Fe) or aluminum (Al) compounds will lead to a decrease of the P pools in soil that are most vulnerable to leaching, and decreases the risk of losses of dissolved P. Examples of materials with a high content of Al and/or Fe are red mud, a waste product from the production of Al [1], and biosolids like municipal sewage sludge [2], alum treated poultry litter [3,8], and drinking water treatment residual [4].

In the past, research on the use of Al/Fe-rich compounds started with the aim of increasing the efficiency of P application and reduce P loss from sandy soils in Australia [5] or to reduce P loss from peat soils in Germany [1]. Later, research was done on the possibilities that applied waste products could immobilize soil P and thus reduce yield [6,7], so immobilization was seen as a negative side-effect. More recent research is focused on immobilizing soil P, especially in over-fertilized soils [2,3,8]. The use of Fe/Al-salts has often the purpose of purification, in order to separate solid and liquid fractions, e.g. for municipal water treatment [2], production of drinking water from ground or surface water [4], or separation of liquid and solid fraction of animal manure [9, link to factsheet]. Also, Fe is used for binding volatile and corrosive sulfur compounds that are formed during anaerobic biogas production from animal manure or biomass [10,11]. Since rest products of these processes are often nutrient rich their agricultural use is common in many countries. Although the impact of the application of these rest products on P losses from soils should receive attention, there is reason to believe that, on a short term, in Europe this type of temporarily binding of soil P will not be seen as a sustainable solution.

### *Applicability*

The measure will be most effective when applied on soils with a very high P availability in the top soil. On these soils vegetative mining of P [12] may take a very long period so this option is less applicable. No specific skills or technical equipment is needed, other than for application of solid materials like sludge. Application of Fe/Al-rich material to buffer zones could be investigated.

### *Effectiveness, including certainty*

When transport of dissolved P via runoff or via preferential flow is important on a site, the effectiveness can be high; this can be the case with artificially drained soils or soils with a high ground water table. When transport of particulate P is important erosion control [see link] must also be applied. However, it can be assumed that the bioavailability of eroded particle bound P is also reduced by addition of immobilizing agents; further investigations are needed to evaluate this assumption. When Fe is the main component in the immobilizing material remobilization can occur in surface water under anaerobic conditions.

### *Time frame*

A reduction in mobile P forms can take place within months after application.

### *Environmental side-effects / pollution swapping*

Besides Fe/Al some biosolids can contain large amounts of heavy metals. In that case, plant availability of these metals must be monitored.

### *Relevance, potential for targeting, administrative handling, control*

The option can be relevant for all fields where P status is far above optimal for crop growth, especially when P export due to leaching to ground water or surface water is significant. Control can be done on a decrease in easily extractable P forms.

### *Costs: investment, labor*

No additional labor costs or investments are known other than costs of spreading a material and incorporating it into the soil layer where immobilization is wanted.

### *References*

- [1] Scheffer, B., R. Bartels, and J. Blankenburg. 1985. Möglichkeiten der Verwertung von Rotschlamm und Grünsalz auf Hochmoorböden. *Z. Pflanzenernaehr. Bodenk.* 148:527-535.
- [2] Maguire, R.O., J.T. Sims, and F.J. Coale. 2000. Phosphorus solubility in biosolids-amended farm soils in the Mid-Atlantic region of the USA. *J. Environ. Qual.* 29:1225-1233.
- [3] Moore, P.A. Jr., and D.R. Edwards. 2005. Long-term effects of poultry litter, alum-treated litter, and ammonium nitrate on aluminum availability in soils. *J. Environ. Qual.* 34:2104-2111.
- [4] Dayton, E.A., and N.T. Basta. 2005. Use of drinking water treatment residuals as a potential best management practice to reduce phosphorus risk index scores. *J. Environ. Qual.* 34:2112-2117.
- [5] Barrow, N.J. 1982. Possibility of using caustic residue from bauxite for improving the chemical and physical-properties of sandy soils. *Aust. J. Agric. Res.* 33:275-285.
- [6] Adriano, D.C., T.A. Woodford, and T.G. Ciravolo. 1978. Growth and elemental composition of corn and bean seedlings as influenced by soil application of coal ash. *J. Environ. Qual.* 7:416-421.
- [7] Cox, A.E., J.J. Camberato, and B.R. Smith. 1997. Phosphate availability and inorganic transformation in an alum sludge-affected soil. *J. Environ. Qual.* 26:1393-1398.
- [8] Smith, D.R., P.A. Jr. Moore, D.M. Miles, B.E. Haggard, and T.C. Daniel. 2004. Decreasing phosphorus runoff losses from land-applied poultry litter with dietary modifications and alum addition. *J. Environ. Qual.* 33:2210-2216.
- [9] Krumpelman, B.W., T.C. Daniel, E.G. Edwards, R.W. McNew, and D.M. Miller. 2005. Optimum coagulant and flocculant concentrations for solids and phosphorus removal from pre-screened flushed dairy manure. *Appl. Eng. in Agric.* 21:127-135.
- [10] Annika, F., T. Andersson, A. Karlsson, and B.H. Svensson. 2004. Occurrence and abatement of volatile sulfur compounds during biogas production. *J. Air Waste Manag. Assoc.* 54:855-861.
- [11] Firer, D., E. Friedler, and O. Lahav. 2008. Control of sulfide in sewer systems by dosage of iron salts: Comparison between theoretical and experimental results, and practical implications. *Sci. Total Environ.* 392:145-156.
- [12] Chardon, W.J. 2008. Vegetative mining. Factsheet COST 869. [www.cost869.alterra.nl/fs/FS\\_mining.pdf](http://www.cost869.alterra.nl/fs/FS_mining.pdf)